PUERARIN 6"-O-β-APIOFURANOSIDE, A C-GLYCOSYLISOFLAVONE O-GLYCOSIDE FROM PUERARIA MIRIFICA

John L. Ingham*, Kenneth R. Markham†, Stanley Z. Dziedzic‡§ and Gerald S. Pope|

*Department of Botany, Plant Science Laboratories, University of Reading, Whiteknights, P.O. Box 221, Reading RG6 2AS, U.K.; †Chemistry Division, D.S.I.R., Petone, New Zealand; †Department of Food Science, Food Studies Building, University of Reading, Whiteknights, P.O. Box 226, Reading RG6 2AP, U.K.; ||Department of Physiology, National Institute for Research in Dairying, Shinfield, Reading RG2 9AT, U.K.

(Received 26 November 1985)

Key Word Index—Pueraria mirifica; Leguminosae—Papilionoideae; isoflavonoids; isoflavone C-glycosides; puerarin; puerarin 6"-O- β -apiofuranoside; ¹H and ¹³C NMR; gated spin echo (GASPE).

Abstract—In addition to puerarin (7,4'-dihydroxyisoflavone 8-C- β -glucopyranoside), the air-dried tuberous roots of *Pueraria mirifica* have been found to contain a second, previously unreported, isoflavone C-glycoside. This new compound (mirificin), which has now been identified by chemical and spectroscopic (UV, ¹H NMR, ¹³C NMR including GASPE) procedures as puerarin 6"-O- β -apiofuranoside is the first O"-glycoside of an isoflavone C-glycoside to be discovered in nature. Mirificin contains a rare $1 \rightarrow 6$ interglycosidic linkage between apiose and the glucose unit which is unique in flavonoids. It is proposed that $1 \rightarrow 2$ and $1 \rightarrow 6$ linked apioglucosides can be distinguished by ¹H NMR spectroscopy in the same manner as used for the equivalent rhamnoglucosides.

INTRODUCTION

Extracts of *Pueraria* roots (Leguminosae-Papilionoideae; tribe Phaseoleae) have yielded various isoflavonoids [1], including a series of novel isoflavone C-glucosides of which puerarin (7,4'-dihydroxyisoflavone 8-C- β -glucopyranoside, 1) is probably the best known example. Puerarin was first obtained from the roots of Chinese and Japanese *Pueraria* species (*P. thunbergiana* = *P. lobata*, *P. pseudo-hirsuta* and *P. thomsonii*) [2, 3] although it has since been found in *P. tuberosa* roots [4], and in callus tissue derived from the stem of *P. thunbergiana* [5]. Four other related isoflavone C-glycosides, namely 3'-hydroxy and 3'-methoxypuerarin [6], the partially identified diglycoside 'puerarin xyloside' [2, 3], and 4',6"-di-O-acetyl-puerarin [4] have also been detected in *Pueraria* roots.

Our interest in P. mirifica Airy Shaw & Suvatabandhu, a woody climber indigenous to parts of northern Thailand, arises from the discovery of a potent oestrogen (miroestrol, 4) in its tuberous roots [7, 8]. Although miroestrol does not possess the 1,2-diphenylpropane skeleton common to all isoflavonoids [1], its A- and Brings closely resemble the corresponding A- and C-rings of daidzein (7,4'-dihydroxyisoflavone, 3), an isoflavone exhibiting weak oestrogenic activity [9]. This suggested that P. mirifica might also be a source of daidzein-related isoflavones, a possibility strengthened by the claim that root extracts contain other oestrogens apart from miroestrol itself [10]. In this paper we describe the isolation of puerarin (1) and a novel isoflavone Cglycoside (mirificin), similarly based on the aglycone daidzein, from P. mirifica roots. Using chemical and

spectroscopic procedures, mirificin has been identified as puerarin $6''-O-\beta$ -apiofuranoside (2).

RESULTS AND DISCUSSION

Silica gel TLC (CHCl₃-MeOH, 20:1) of a P. mirifica root extract yielded numerous bands fluorescing blue under long wavelength (ca 365 nm) UV light. However, our attention was drawn particularly to the lowest region of the chromatogram $(R_f 0.00-0.05)$ which fluoresced pale blue, changing to bright blue upon fuming with NH, vapour, a feature often associated with isoflavones lacking oxygenation at C-5 [11]. Further purification of this nearorigin material in chloroform-methanol-water (20:10:1) gave two compounds (1 at R_c 0.53 and 2 at R_c 0.37), both of which exhibited UV maxima similar to those of authentic daidzein (3). Upon treatment with boiling 2 N HCl [11] compound 1 was unaffected, whereas 2 readily yielded 1 and a sugar that was not glucose, galactose, rhamnose, xylose or arabinose. These data, together with its R_f values (TLC) on cellulose plates developed in TBA (0.69) and 15% acetic acid (0.72), suggested that 1 was an isoflavone C-glycoside, and that 2 was an O-glycoside of 1 with the additional sugar(s) probably being attached to the C-glycosyl moiety.

The ¹H NMR spectrum of 1 revealed a pattern of aromatic proton signals resembling that of daidzein (Table 1) except for the H-8 doublet at δ 6.83 which was absent, and the lack of *meta*-coupling in H-6. Thus 1 was provisionally formulated as an 8-C-glycosylated derivative of daidzein, a view later confirmed by ¹³C NMR spectroscopy (Table 2). The ¹³C NMR data also established the C-linked sugar as β -glucopyranose, evidence for a β -linkage being additionally provided by the 9.3 Hz, H-1"/H-2" coupling constant in the ¹H NMR

Short Reports 1773

Table 1.	¹ H NMR d	a (ô value	s) for daidzein ((3), puerarin ((1) and mirificin (2)*
----------	----------------------	------------	-------------------	-----------------	------------------------

	Daidzein†	Puerarin	Mirificin (2)	
Proton	(3)	(1)		
H-2	8.22	8.21	8.19	
	(1H, s)	(1H, s)	(1H, s)	
H-5	7.94	7.82	7.78	
	(1H, d, J = 9.0)	(1H, d, J = 8.8)	(1H, d, J = 8.8)	
H-6	6.90	6.85	6.80	
	(1H, dd, J = 9.0 and 2.0)	(1H, d, J = 8.8)	(1H, d, J = 8.8)	
H-8	6.83	<u> </u>	_	
	(1H, d, J = 2.0)			
H-2',6'	7.36	7.40	7.39	
•	(2H, d, J = 9.0)	(2H, d, J = 8.5)	(2H, d, J = 8.6)	
H-3',5'	6.79	6.80	6.80	
•	(2H, d, J = 9.0)	(2H, d, J = 8.5)	(2H, d, J = 8.6)	
	•	•	(4.761	
Sugar H-1	_	4.82	(1H, d, J = 10.8)	
_		(1H, d, J = 9.3)	4.79	
		, ,	(1H, d, J = 3.0)	

^{*}Spectra were determined in DMSO- d_6 at 80 MHz (1), 100 MHz (3) and 200 MHz (2). Coupling constants (J in Hz) are given in parentheses.

spectrum. Isoflavone 1 is therefore defined as daidzein 8-C- β -glucopyranoside (puerarin). Samples of puerarin derived from P. mirifica root and stem callus of P. thunbergiana [5] proved to be inseparable when co-chromatographed (TLC) on cellulose plates in 15% acetic acid (R_f 0.72) and TBA (R_f 0.69).

Apart from an extra H-1 sugar signal, the ¹H NMR spectrum of isoflavone 2 appeared very similar to that of puerarin (Table 1). The C-linked glucose H-1 signal was evident as a doublet at $\delta 4.76$ (J = 10.8 Hz), and the second H-1 resonance at $\delta 4.79$ (J = 3 Hz) was therefore assigned to the hydrolysable O-linked sugar. In the ¹³C NMR spectrum, this sugar residue gave a pattern of five signals with chemical shift values closely resembling those previously published for glycosides containing β -apiose [12, 13], an unusual branched chain pentafuranose. The presence of apiose in 2 was confirmed in two ways. First, by a gated spin echo (GASPE) [14] 13C NMR experiment which defined the three signals at δ 78.9, 73.4 and 63.4 as representing quaternary or methylene carbons, and the remaining signals (δ 109.1 and 75.9) as methine carbons. This experiment also confirmed the assignments shown for these signals in Table 2. Secondly, apiose was identified in the hydrolysate of 2 by paper chromatographic comparison with authentic apiose derived from the flavone apiin [15]. Attachment of apiose, via an O-link, to the glucose residue rather than to the isoflavone nucleus at C-7 or C-4' was apparent from the UV shift observed with sodium acetate (C-7 OH [11]), and from the B-ring proton δ values (Table 1) which were essentially identical with those obtained for puerarin and daidzein. Lastly, the interglycosidic linkage in 2 was clearly defined as 1 -> 6 by the 13C NMR spectrum. Thus the glucose C-6 signal was shifted downfield by 7 ppm relative to that of puerarin (Table 2), whereas the C-5

signal exhibited the expected upfield shift. The chemical shift of the apiose C-1 was essentially the same as previously reported for C-1 linked apiosides [12, 13]. Compound 2 is therefore 7,4'-dihydroxy-8-C- β -glucopyranosylisoflavone 6"-O- β -apiofuranoside (puerarin 6"-O- β -apiofuranoside), a new natural product for which we propose the name mirificin.

To date, puerarin has only been associated with the genus Pueraria [1], and the present work extends its known occurrence to yet another Pueraria species. Both flavonoid and isoflavonoid glycosides containing apiose are rare in nature but have been isolated on two previous occasions from leguminous plants. Thus, in 1967 Malhotra et al. [16] extracted the isoflavone biochanin A 7-O-apiosyl(1 \rightarrow 2) glucoside from the roots of Dalbergia lanceolaria, and more recently the flavanone liquiritigenin 4'-0-apiosyl(1 \rightarrow 2)glucoside was obtained from licorice (Glycyrrhiza uralensis) roots [13]. However, whilst an Olinked $1 \rightarrow 6$ apinglucoside of a xanthone has already been discovered in Polygala caudata (Polygalaceae) [17], mirificin from P. mirifica is apparently the first C-linked apiosyl(1 → 6)glucoside of any kind to be recognised as a natural product.

The present work has also revealed an interesting analogy between the sugar H-1 chemical shift data of rhamnosyl($1 \rightarrow 6$)glucosides (rutinosides)/rhamnosyl($1 \rightarrow 2$)glucosides (neohesperidosides) and the equivalent apiosylglucosides. With the two rhamnosylglucosides, it has been observed that although the glucose H-1 shifts are similar, the rhamnose H-1 signal in neohesperidosides occurs at about 0.5 ppm downfield from its position in rutinosides. This difference is considered to be a reliable method of distinguishing between these two most common rhamnosylglucosides [11]. Significantly, the apiosylglucosides closely parallel this situation, with the

[†]Data for daidzein (3) are taken from ref. [5].

[‡] Upper δ value refers to H-1 of the C-linked sugar (glucose); lower δ value is for H-1 of the O-linked sugar (apiose).

1774 Short Reports

Table 2. ¹³C NMR data (δ values) for daidzein (3), puerarin (1), mirificin (2) and apiose*

	Daidzein (3)†	Puerarin (1)	Mirificin (2)	Apiose (apiin)†
Isoflavone				
C-2	152.9	152.0	151.8	
C-3	123.6 ^b	123.1°	123.2 ^b	_
C-4	175.0	174.8	174.8	
C-5	127.4	125.8	125.7	
C-6	115.3	114.3	115.1	
C-7	162.7	166.4	167.1	
C-8	102.2	111.9	111.8	
C-9	157.3ª	157.1*	157.2	
C-10	116.7	117.4	117.6	
C-1'	122.7 ^b	122.9°	122.8 ^b	
C-2'	130.2	130.1	130.1	
C-3'	115.1	115.1	115.1	
C-4'	157.7ª	157.3ª	157.2	
C-5'	115.1	115.1	115.1	
C-6'	130.2	130.1	130.1	
C-sugar				
C-1"	_	74.1	74.2	
C-2"	_	70.8 ^b	70.9	
C-3*	_	79.2	78.9	
C-4"	_	70.5 ^b	70.9	
C-5"		81.6	80.1	_
C-6"		61.3	68.4	
O-sugar				
C-1‴		_	109.1	109.0
C-2‴	_		75.9	76.5
C-3‴	-	_	78.9	79.1
C-4"		_	73.4	74.0
C- 5 ‴		_	63.4	64.4

^{*}Spectra were determined in DMSO- d_6 (30°) at 20 MHz. Assignments bearing the same superscript (a, b or c) in any one spectrum may be reversed.

apiose H-1 signal appearing at δ 4.79 in the (1 \rightarrow 6)-linked diglycoside (Table 1), and at δ 5.38 in the (1 \rightarrow 2)-linked diglycoside [13]. It is likely, therefore, that ¹H NMR data for apiosylglucosides has diagnostic value similar to that previously demonstrated for rhamnosylglucosides.

EXPERIMENTAL

Plant material. Tuberous roots of Pueraria mirifica Airy Shaw & Suvatabandhu [18] (collected by Dr. S. Bartlett from the Doi Suthep area near Chiangmai in northern Thailand) were sliced and air-dried before shipment to the U.K. Upon receipt, the root material was stored in sealed glass jars away from sunlight until required for extraction.

Extraction, isolation and purification of puerarin (1) and mirificin (2). The powdered root (870 g) was first stirred with boiling 95% MeOH (4.3 l.) for 30 min. After cooling and filtering, the residue was extracted twice more with a mixture of 95% MeOH (2 × 3.5 l.) and $\rm H_2O$ (2 × 870 ml). The combined filtrates (ca 15 l.) were taken to dryness in vacuo (40°) and the residue was then stirred with 900 ml of 95% MeOH. The resulting suspension was filtered to remove insoluble waxy material, and the filtrate

was then coned in vacuo to ca 450 ml before being filtered again. An equal vol. of Et₂O was added to the filtrate, and after standing overnight at room temp, the clear MeOH-Et₂O soln was decanted. The gummy ppt was washed with more Et₂O (2 × ca 100 ml) and these washings were added to the MeOH-Et₂O soln which was immediately reduced to dryness. The residue thus obtained was dissolved in MeOH (130 ml) and gently shaken with 300 ml of Et₂O. Upon settling, the clear supernatant was decanted and the remaining solid was re-dissolved in MeOH (200 ml). A final quantity of Et₂O (300 ml) was added to this MeOH soln giving a ppt which was removed and air-dried. The resulting brownish cake was then treated as follows: (a) a small portion (ca 100 mg) was dissolved in boiling MeOH (ca 10 ml) and applied to a number of pre-coated preparative silica gel thinlayer plates (Merck, glass-backed, F-254, layer thickness 0.5 mm) which were developed in CHCl₃-MeOH (20:1) to afford various bands fluorescing blue under long wavelength (ca 365 nm) UV light. The zone of silica gel extending from the origin to $caR_10.05$ was removed and its isoflavonoid components were eluted with MeOH (50 ml). Concn of the cluate in vacuo (40°) to ca 5 ml afforded a yellow-brown soln which was chromatographed (silica gel TLC, layer thickness 0.25 mm) in CHCl₃-MeOH-H₂O

[†]Assignments are based on data given in ref. [12].

Short Reports 1775

(20:10:1) to give puerarin (1) and mirificin (2) at R_f 0.53 and 0.37 respectively. (b) A larger quantity (ca 3.5 g) of the finely powdered Et₂O-precipitated material was applied to a dry pack silica gel column (Merck, Type 7734, 300 g) prior to elution with CHCl₃-MeOH-H₂O (20:10:1). Eluant fractions (10 ml each) were monitored by silica gel TLC, with puerarin and mirificin being detected by their characteristic pale blue fluorescence under 365 nm UV light. Fractions containing mainly mirificin were combined and reduced to dryness in vacuo (45°). The residue was then chromatographed (silica gel TLC) in CHCl₃-MeOH-H₂O (20:10:1) to afford, after elution of the appropriate silica gel band with Me₂CO, ca 14 mg of pure mirificin. This material was principally used for the spectroscopic (¹H and ¹³C NMR) studies.

Puerarin (7,4'-dihydroxyisoflavone 8-C-β-glucopyranoside, 1). Appearance on silica gel TLC plates viewed under long wavelength UV light before and after fuming with NH₃ vapour, pale blue and bright blue respectively. Colour with diazotized p-nitroaniline reagent, orange. UV λ_{max}^{MeOH} nm (% intensity): 206 (100%), 243 sh (83%), 251 (88%), 263 sh (77%), 309 (30%); λ_{max}^{MeOH-NaOH} nm: 213, 265, 292 sh, 336; λ_{max}^{MeOH-NaOAc} nm: 260, 342 (addition of H₃BO₃ regenerated the MeOH spectrum). Comparative UV maxima recorded for synthetic daidzein (7,4'-dihydroxyisoflavone, 3) were: λ_{max}^{MeOH} nm: 211, 242 sh, 250, 263 sh, 306; λ_{max}^{MeOH+NaOH} nm: 215, 260, 291 sh, 331; λ_{max}^{MeOH+NaOAc} nm: 255, 272 sh, 337 (addition of H₃BO₃ regenerated the MeOH spectrum). ¹H and ¹³C NMR data for puerarin are shown in Tables 1 and 2 respectively.

Mirificin (7,4'-dihydroxy-8-C-\beta-glucopyranosylisoflavone 6"-O-β-apiofuranoside, 2). Fluorescence characteristics and colour with diazotized p-nitroaniline reagent as for puerarin (1). UV λ_{max}^{MeOH} nm (% intensity): 208 (98%), 242 sh (90%), 252 (100%), 263 sh (88%), 308 (33%); \(\lambda_{\text{max}}^{\text{McOH-NaOH}}\) nm: 210, 263, 293 sh, 336; 1 MeOH-NaOAc nm: 259, 340 (addition of H₃BO₃ regenerated the MeOH spectrum). For ¹H and ¹³C NMR data see Tables 1 and 2 respectively. Mirificin was hydrolysed with boiling 2 N HCl according to standard practice [11]. Apiose was detected in the hydrolysate PC in bv BuOH- C_6H_6 -pyridine- H_2O (5:1:3:3). Relative R_7 values were: galactose 0.27, glucose 0.32, arabinose 0.35, xylose 0.41, apiose 0.49 and rhamnose 0.53. The hydrolysate also yielded a compound inseparable from puerarin (1) by silica gel TLC in CHCl₃-MeOH-H₂O (20:10:1, R_f 0.53).

Acknowledgements—We thank Dr. H. Itokawa (Tokyo College of Pharmacy, Japan) for a sample of puerarin, and Dr. H. Wong (Chemistry Division, D.S.I.R., Petone, New Zealand) for determining the ¹H and ¹³C NMR spectra.

REFERENCES

- 1. Ingham, J. L. (1983) Fortschr. Chem. Org. Naturst. 43, 1.
- Shibata, S., Murakami, T., Nishikawa, Y. and Harada, M. (1959) Chem. Pharm. Bull. Tokyo 7, 134.
- Murakami, T., Nishikawa, Y. and Ando, T. (1960) Chem. Pharm. Bull. Tokyo 8, 688.
- Bhutani, S. P., Chibber, S. S. and Seshadri, T. R. (1969) Indian J. Chem. 7, 210.
- Takeya, K. and Itokawa, H. (1982) Chem. Pharm. Bull. Tokyo 30, 1496.
- Natori, S., Ikegawa, N. and Suzuki, M. (1977) Experimental Methods of Organic Natural Products, p. 470. Kodansha, Tokyo.
- 7. Cain, J. C. (1960) Nature 188, 774.
- 8. Bounds, D. G. and Pope, G. S. (1960) J. Chem. Soc. 3696.
- Bickoff, E. M., Livingston, A. L., Hendrickson, A. P. and Booth, A. N. (1962) J. Agric. Food Chem. 10, 410.
- 10. Jones, H. E. H. and Pope, G. S. (1961) J. Endocrinol. 22, 303.
- Mabry, T. J., Markham, K. R. and Thomas, M. B. (1970) The Systematic Identification of Flavonoids. Springer, New York.
- Markham, K. R., Chari, V. M. and Mabry, T. J. (1982) in The Flavonoids—Advances in Research (Harborne, J. B. and Mabry, T. J., eds) p. 19. Chapman & Hall, London.
- 13. Yahara, S. and Nishioka, I. (1984) Phytochemistry 23, 2108.
- Cookson, D. J. and Smith, B. E. (1981) Org. Magn. Reson. 16, 111.
- Gupta, S. R. and Seshadri, T. R. (1952) Proc. Indian Acad. Sci. 35A, 242.
- Malhotra, A., Murti, V. V. S. and Seshadri, T. R. (1967) Tetrahedron 23, 405.
- Pan, M. and Mao, Q. (1984) Yaoxue Xuebao 19, 899 (Chem. Abstr. 103, 19852w, 1985).
- Kashemsanta, L., Suvatabandhu, K. and Airy Shaw, H. K. (1952) Kew Bull. 549.